



UNIVERSITY OF ABERDEEN

THE NATURE OF LIGHT

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EG 40GC Optical Engineering
LECTURE NOTES

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The Nature of Light

- If the laser is at the focus of optical engineering,
 - **then light is at its heart**
- The basis of optical engineering is
 - **that light is the prime carrier of information**
- The optical engineer must be familiar with, and have an intuitive understanding of,
 - **the unique character of light**
 - **its origins in the atomic structure of matter.**
- We need to outline the nature of light
 - **to enable us to understand principal optical phenomena**
 - **to appreciate the unique properties of the laser.**

Waves & Photons

- Light is not the easiest of natural phenomena to describe.
- Light has an ambivalent nature that requires
 - ***two models of behaviour*** for a satisfactory description of its properties.
- The electromagnetic wave model
- The quantum electrodynamic (photon) model
- These models, although seemingly contradictory, are complementary
 - **Both models are needed to give a complete picture of the nature of light and to emphasise its role as an information carrier.**

The electromagnetic wave model

- Describes light as
 - a **continuous transfer of energy** through space
 - by a combination of waves of electric & magnetic fields
 - **describes propagation of light**
 - **helps to describe phenomenon like**
 - black-body radiation
 - diffraction
 - interference
 - polarisation

The quantum electrodynamic (photon) model

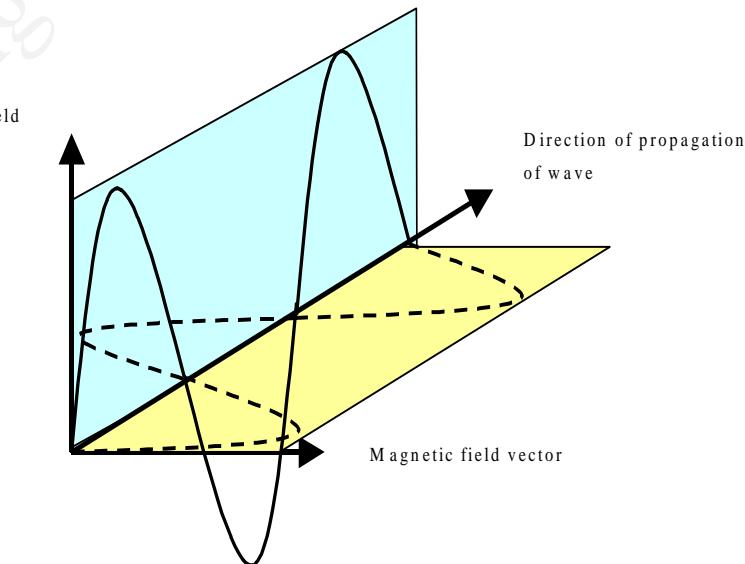
- Describes light in terms of massless elementary particles,
 - known as photons
 - can be thought of as localised pulses of energy
- Model is needed to describe interactions between light & matter
 - helps us to describe absorption and emission of light from materials

The Electromagnetic Wave Model of Light

- The theory of electromagnetic radiation was proposed by Scottish scientist James Clerk Maxwell
- It describes the continuous transfer of energy through space by the propagation of electro-magnetic waves.
 - Mutual interaction of two vector fields, an electric field ξ and a magnetic field \mathcal{H}
- In any dynamic situation, a propagating ξ -field will be accompanied by a corresponding \mathcal{H} -field propagating everywhere at right angles to it
 - Radiation propagates in a direction that is mutually perpendicular to both ξ - and \mathcal{H} -fields.

The Electromagnetic Wave Model of Light

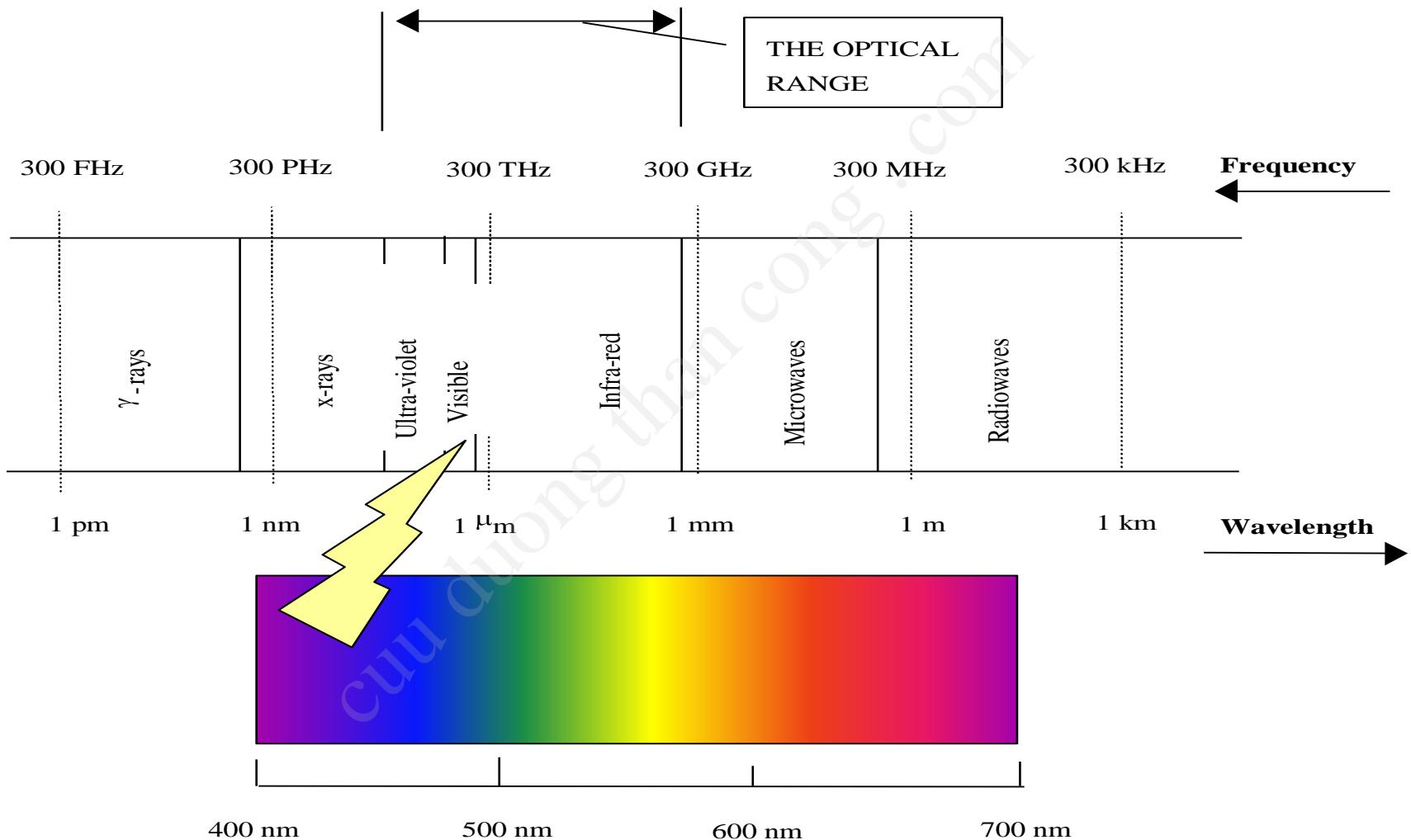
- The ξ - & \mathcal{H} -fields have *no component* of vibration in the direction of propagation
 - **no longitudinal component**
 - **exclusively transverse in nature**
- Convention in optics describes the behaviour of e-m field in terms of the oscillating electric vector
 - **ξ -field is always accompanied by an associated \mathcal{H} -field.**
 - **knowing one allows us to extract the other**



The Electro-Magnetic Spectrum

- Lightwaves, microwaves, radiowaves, X-rays and gamma rays are all one and the same thing
 - electromagnetic radiation.
 - they differ only in their characteristic frequency.
- The complete *electromagnetic spectrum* stretches continuously from γ -rays to radiowaves
- The optical range is usually considered to stretch from 200 nm (far ultra-violet) to about 20 μm (far infrared)
- Visible light, stretches from about 400 nm at the violet end to around 700 nm at the red end,
 - **forms only a small part of the e-m spectrum.**

The e-m Spectrum

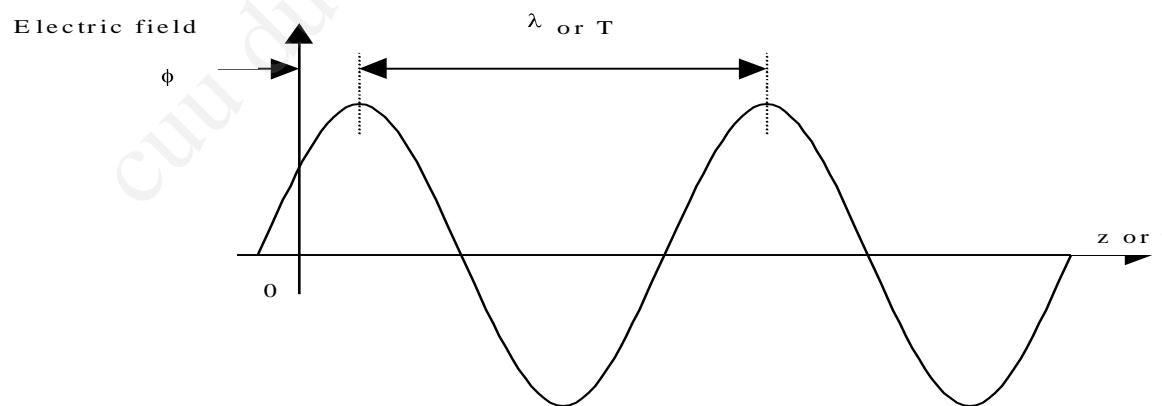


Plane Waves - 1

- A complete, formalised description of the propagation of light through free space and the energy transferred by it, requires that
 - **we treat electro-magnetic (e-m) radiation as a vector field**
 - **specify in 3-d space, in terms of its magnitude and direction.**
- For most purposes we may represent a travelling light wave as
 - **a *one-dimensional, scalar wave*, propagating in a given direction.**
 - **Such a wave is known as a plane wave**
 - **all the surfaces upon which the wave has equal phase are parallel to each other and perpendicular to the direction of propagation.**
 - **The planes can be considered as the wavefronts of the wave and usually represent the peak amplitude of the wave**
 - **separated by the wavelength**
- We need only describe such a wave in terms of either the electric field or the magnetic field.
 - **Both are not necessary, since we can always extract one from the other.**

Plane Waves - 2

- Conventionally, amplitude of electric field vector, ξ , describes a plane wave of angular frequency ω and wave vector, k , propagating in the $+z$ direction
 - $\xi = \xi_0 \cos(\omega t - kz - \phi)$
 - ϕ is the initial phase of the wave which depends on the chosen location of the origin
 - t and z are the respective time and space co-ordinates
 - ξ_0 is the peak amplitude of the wave
 - ω is angular frequency
 - $\omega = 2\pi\nu$ [rad s⁻¹] ν is linear frequency of light
- Wavefronts correspond to the peaks of the sine wave
 - They are perpendicular (normal) to direction of propagation

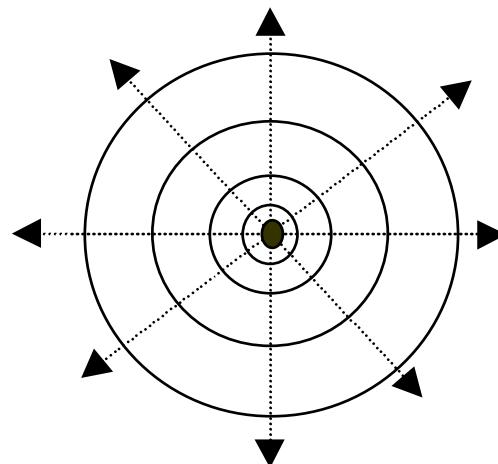


Plane Waves - 3

- The propagation constant is
 - $k = 2\pi/\lambda$ [m⁻¹]
 - λ is the wavelength.
- The velocity of propagation of the light wave in a vacuum,
 - $c = v\lambda$ [m s⁻¹]
- The velocity of the wave in any medium is related to its free-space velocity
 - $v = c/n = v(\lambda/n)$
- The refractive index itself is related to the relative magnetic and electric constants, ϵ_r and μ_r
 - $n = c/v = \sqrt{(\epsilon_r \cdot \mu_r)}/(\epsilon_0 \cdot \mu_0)$
- The constants, ϵ_0 and μ_0 are:
 - $\epsilon_0 = 8.87 \times 10^{-12} \text{ F m}^{-1}$ **primary electric constant of free space**
 - $\mu_0 = 4 \times 10^{-7} \text{ H m}^{-1}$ **primary magnetic constant of free space**
- The velocity of propagation of an electromagnetic wave in free space (i.e. a vacuum)
 - $c = \sqrt{(\epsilon_0 \mu_0)} = 2.998 \times 10^8 \text{ m s}^{-1} \approx 3 \times 10^8 \text{ m s}^{-1}$

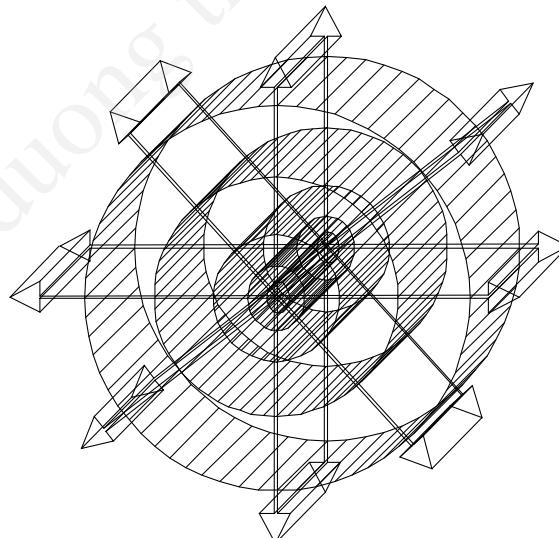
Spherical Waves

- Some sources of light generate waves that spread radially outwards equally in all directions from a common point of emission.
 - **wavefronts expand outwards in concentric spheres with ever increasing diameters**
 - **known as** isotropic point sources
- The spherical wave decreases in amplitude as it propagates
 - $\xi = (\xi_0/r) \cos(\omega t - kr)$
 - **at large distances from the point source, the spherical wave can be approximated by a plane wave.**



Cylindrical Waves

- The electric field distribution of a cylindrical wave is given as
 - $\xi = (\xi_0/r^{1/2}) \cos(\omega t - kr)$
- Cylindrical waves are good approximations to the source profile emitted when a plane wave is incident on a long slit or the radiation profile from a linear flashtube.

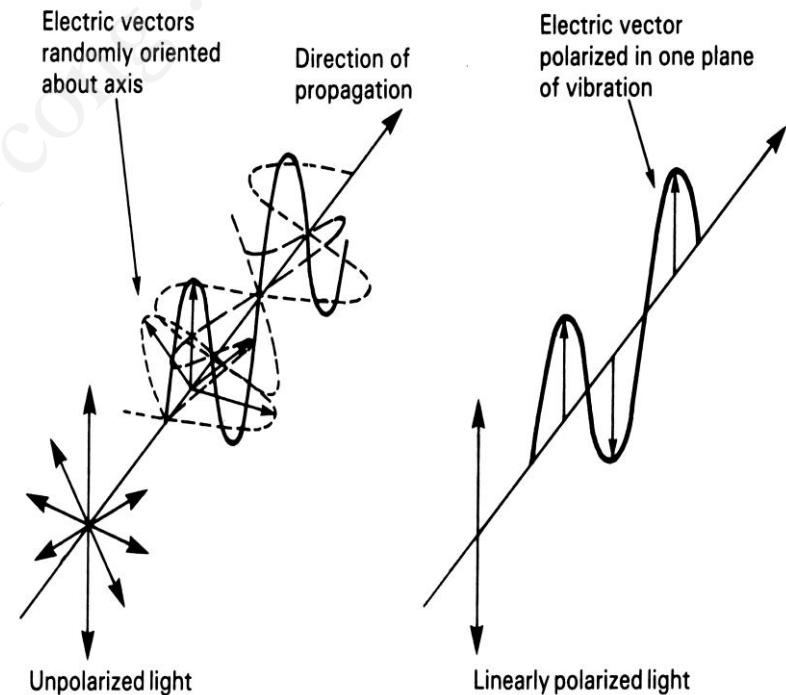


Irradiance of Light

- The **irradiance** of an e-m wave is the *power per unit area* that is incident on a surface
 - $E = \rho c = <\xi^2>/Z_0$ [W m⁻²]
 - ρ is the **radiation density** (energy per unit volume)
 - Z_0 is the **characteristic impedance of free space**,
 - $Z_0 = \xi_x/\mathcal{H}_y = (\mu_0/\epsilon_0)^{1/2} = 377 \Omega$
- Light can also exert pressure on a surface
 - **Average radiation pressure is defined in terms of irradiance**
 - $P_{rad} = E/c$ [J m⁻³]

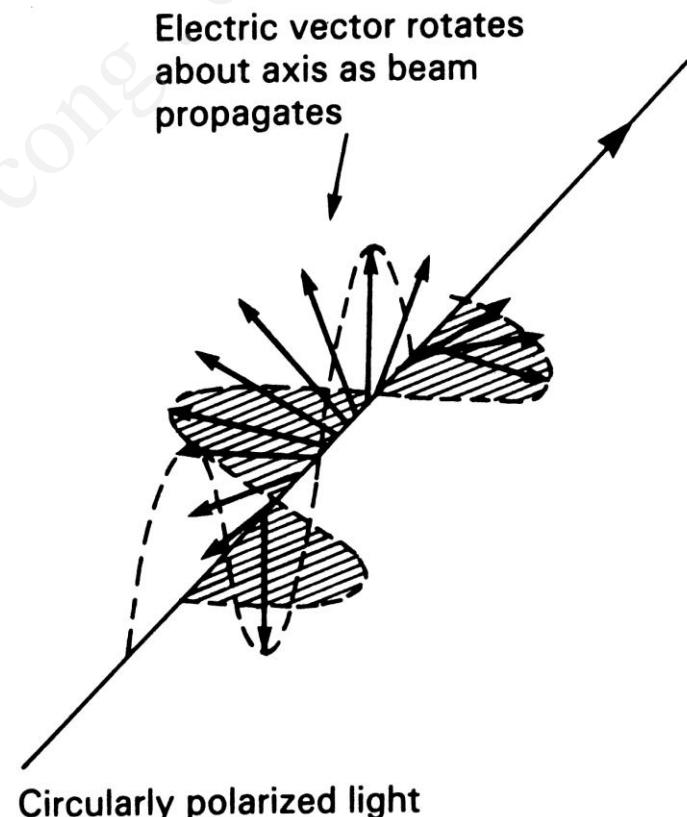
Polarisation of Light Waves - Linear

- A propagating light wave may be expressed uniquely in terms of its ξ -vector.
 - For most sources of light,
 - the radiant emission will be composed of many waves
 - each with their ξ -vectors randomly orientated wrt each other.
 - Beam is unpolarised or randomly polarised.
- In some cases waves are constrained to oscillate in preferential planes
 - Beam is polarised
- When the light wave oscillates in a single plane of the electric vector
 - Beam is linearly or plane polarised
 - Normally the plane of vibration of ξ -vector will be specified e.g. vertical



Polarisation of Light Waves – Circular & Elliptical

- More complex situation occurs when the plane of polarisation continually rotates perpendicularly to the direction of propagation as the wave propagates.
 - Beam is **circularly polarised**
 - **Two senses: *left and right* polarisation, denote the direction of rotation of the polarisation plane.**
- In some cases, the amplitude of the polarisation vector also changes with the rotation as the wave propagates
 - Beam is **elliptically polarised**



The Photon Model of Light

- The propagation of light through space may be described in terms of a travelling wave motion
 - **emission & absorption of light by materials is not easy to describe on this basis**
- Whether or not light existed as a stream of particles or as a wave motion was the subject of heated debate for hundreds of years
- There are some phenomena which defy all attempts to describe them on the basis of a classical wave model of light propagation
 - **emission of light from gases and solids**
 - **black body radiation**
 - **the photoelectric effect**
- It is necessary to invoke the idea of light being emitted in tiny pulses, or **quanta** of energy
 - **not as a continuous distribution of energy.**
 - **the pulses of light are known as photons**
 - **the amount of energy carried by a photon is proportional to the frequency of the radiation**

Energy Imparted by a Photon

- Light can be considered to propagate through space as a stream of vast numbers of “particles”.
 - This particle is not a “hard” lump of well-defined mass, but a massless, quantum of energy
- Each photon carries with it a precisely defined amount of energy which depends only on its characteristic frequency or wavelength
 - The energy of a single photon is
 - $W_{ph} = h\nu_{ph} = hc/\lambda_{ph}$
 - $h = 663 \times 10^{-36} \text{ J s}$ Planck's constant
 - c is the velocity of propagation of the photon in free space

Energy of a Photon

- A particular photon has a wavelength of 600 nm. This photon would be in the red part of the spectrum.
- The energy of a single photon is,
- $W_{ph} = h\nu_{ph} = hc/\lambda_{ph}$
 $= (663 \times 10^{-36} \text{ J s} \times 300 \times 10^6 \text{ m s}^{-1})/(600 \times 10^{-9} \text{ m})$
- $= 332 \times 10^{-21} \text{ J}$
 - **This is a tiny amount of energy!**
- An **electronvolt (eV)** is defined as the energy obtained by an electron when it is accelerated through a potential difference of 1 volt.
 - $1 \text{ eV} = 160 \times 10^{-21} \text{ J}$
- Therefore, the energy of the 600 nm photon, in electronvolts, is
 - $W_{ph} = 332 \times 10^{-21} \text{ J} / 160 \times 10^{-21} \text{ J eV}^{-1} = 2.07 \text{ eV}$
- Visible photons range in energy from 1.74 eV (700 nm) to 3.34 eV (400 nm).

Number of Photons in a Beam of Light

- For a parallel beam of light of 0.5 J energy and 600 nm wavelength, the number of photons contained in this beam is given by
 - $N_{ph} = Q/W_{ph}$
- The energy of a 600 nm photon is 332×10^{-21} J. Thus the number of photons is
 - $N_{ph} = Q/W_{ph}$
 - $= 0.5 \text{ J} / 332 \times 10^{-21} \text{ J}$
 - $= 1.5 \times 10^{18}$
- Total energy or radiant power of a beam of light defines the total number of photons contained within the beam.
- Photon energy gives us the energy of an individual photon.